

DEPARTMENT OF PHYSICS
 OSAKA IMPERIAL UNIVERSITY.

DATE Dec 19

NO. 1955

第六十号

物理院集録

- 1) Metal \rightarrow Crystal Structure \rightarrow Metal Atom \rightarrow Electronic Configuration
- 2) J.C. Slater's Model (Spin Valency)
 - i) Body Centred \rightarrow Exchange Neg.
 - ii) Face Centred \rightarrow Coulomb Force, or Exchange Positive
 - iii) Hexagonal \rightarrow Coulomb Force. (Exchange Small Neg.)
- 3) Inner Cell of d & f
 - i) Coulomb force.
 - ii) Exchange Posi. (Fenomena) \rightarrow d or f electron interaction
 - iii) Resultant Spin, Resultant L of d & f

spin exchange interaction

d & f interaction \rightarrow f & d interaction. (4-77)

(Mn, Co, Ni, Fe, Cr, V, Ti, Zr, Nb, Mo, Tc, Ru, Rh, Pd, Ag, Au, Hg, Pt, Au, Hg, Pb, Bi, Po, At, Rn)

α : Ferro. \boxtimes β : para. \boxtimes γ : para. \square

Ferro \boxtimes 9. 20. 21. 22. 23. 24. Model of d & f interaction...
- 4) Wigner-Seitz's Model (Free Electron Model)

is not quantitative with transition metals.

electron interaction of d & f electrons.

(Slater's v Polar Model is better for d & f .)